# Lect 2- Organic Chem Biomolecules student copy

#### Other organic compounds

Take a cheeseburger.... hamburger, covered with American (yellow) cheese on a hamburger bun... yummy! Now, if you made this cheeseburger with Swiss cheese and put it on slices of rye bread, you'd end up with a "cheeseburger" but one that tasted totally different ... you would notice that the substitutions affected the taste... Chemists make similar changes to organic compounds... these changes produce compounds called

A substituted hydrocarbon has had one or more of its hydrogen atoms or groups of atoms replaced by atoms or groups of atoms of other elements.

alcohol is the name of a family of compounds formed when a hydroxyl (-OH) group one or more hydrogen atoms in a hydrocarbon.

(ex: thanolis produced by sugar fermenting in corn, grains & fruits) Structure challenge:

Alcohols Alcohols Nomenclature: Characterized by the hydroxyl group CH-OH -OH methanol (methyl alcohol)

Isopropyl alcohol: The -OH is on the middle carbon of the 3 carbon chain Propyl alcohol: Has the carbon on the end

a carboxylic acid is formed when a group is replaced by a carboxyl (-COOH) group. (The simplest carboxylic acid is methanic acid or formic acid which is made by ants and is injected into your skin when they bite you)

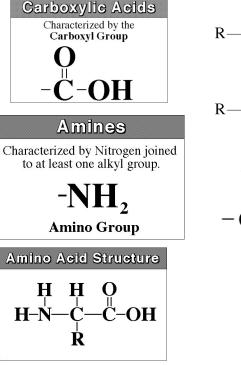
In this group, Nitrogen forms bonds with the carbon and hydrogen. The amine group replaces the hydrogen in the hydrocarbon. Mathylamine is the simplest amine. (EX: novicane in the dentist's office, caffeine in soft drinks... are all hydrocarbons substituted with nitrogen) Example: Ethylmethylamine: CH<sub>3</sub>NHCH<sub>2</sub>CH<sub>3</sub>

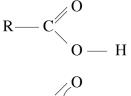
Amino acids have a  $-NH_2$  group along as well as an acid group in it's structure, and is a building block for proteins. They also have both: groups ( a substituted hydrocarbon and more than one chemical group replacing its hydrogen at one time)

Milk, blood muscle, cassette tapes & athletic shoes are all made of organic compounds with called Polymers. Polymers are made up of smaller organic compounds that are linked together. to form new bonds. Polymers are also found in the biological compounds that make up living things.

Biomolecules: Chemistry of Living Systems What are Biomolecules?

- compounds are made by living things
- Also called biochemicals Some are very large Biomolecules are based on the most important element to living organisms:
- There are thousands of different biomolecules but only 4 categories:





Amino



#### Addition Polymerization

The simplest and most widely used addition polymer is polyethylene

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#### <u>3.</u> Made of amino acids Sugar Vitamins & Minerals Starch Cellulose DNA Fats Oils **RNA**

**4 Biomolecule Categories** 

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Sweet Crackers Mini Lab: What happened to the taste of the cracker during the	1) What does the change in taste
time it was in your mouth?	tell you about the molecules in the
	cracker?

# Carbohydrates

Facts       The biomol       Contain 3 elements:	ecules	Two Types of Carbohydrates 1 Examples:
•The word carbohydrate comes f many carbon atoms bonded to	rom the fact that these compounds have (OH) groups.	
•Serve 2 main functions: Source of chemical Part of the	for cells in many living things. material of plants	2 Examples:
•Come in all sizes, from small ri	ngs to long chains	

Simple Carbohydrates: Sugar: Break down \_\_\_\_\_\_ in the body Provide a quick burst of energy or a "sugar rush".\_\_\_\_\_\_ is the most important & simplest sugar on Earth. Used in cells & created by photosynthesis It comes in many forms.

Complex Carbohydrates: \_\_\_\_\_ Long \_\_\_\_\_ of of simpler sugars joined together. They're also called polysaccharides or \_\_\_\_\_ ("poly" = many). \_\_\_\_\_\_ to break down in the body & provide can energy for a longer period of time than regular sugars. This is why marathon runners, cyclists, and most athletes

"carbo-load" the weeks before a big event... to store up energy.

Starches are found in corn, potatoes, wheat, pasta, and many other foods.

Cellulose: Unlike animal cells, \_\_\_\_\_\_cells have a cell wall. This cell wall is a touch protective layer made out of cellulose, a \_\_\_\_\_\_ (or BIG molecule). The cell wall is a large part of vegetables such as lettuce & celery. It is also what gives stems and wood strength. Like starch, cellulose is made from chains of thousands of glucose molecules, but the difference is in how they're linked. Because of this small difference, your body \_\_\_\_\_\_ digest cellulose the same way it can starches & sugars.

Carbohydrates Summary: Carbohydrates are organic compounds in which there are \_\_\_\_\_\_as many \_\_\_\_\_\_atoms as \_\_\_\_\_atoms. Like fats, they contain carbon, hydrogen and oxygen. However, in carbohydrates, the hydrogen and oxygen are present in a ratio of 2 hydrogen atoms to one oxygen atom. This ratio is the same as water. The sugar in blood is called glucose. and has the formula  $C_6H_{12}O_6$ Sucrose  $C_{12}H_{22}O_{11}$  is also a common sugars. \_\_\_\_\_\_\_ is found in honey and grapes. Starches are larger molecules that occur naturally in wheat, rice and corn ( a natural starch food is pasta and bread!!) **DO BrainPop: Carbohydrates** 

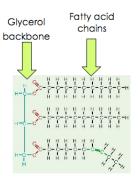
LipidS Facts: Lipids include: \_\_\_\_\_\_ and \_\_\_\_\_ Like carbohydrates, most lipids are also made of just \_\_\_\_\_\_\_ hydrogen, and oxygen. But unlike carbohydrates, fats are more complex and take much longer to break down. So, fats are \_\_\_\_\_\_\_ molecules that plants and animals use to store energy in reserves for longer periods. Plants store energy in \_\_\_\_\_\_\_ like olive oil or peanut oil. An oil is a fat that is \_\_\_\_\_\_\_ at room temperature. Lipids Structure: Fats & oils store energy super-efficiently, 1 gram of fat contains about \_\_\_\_\_\_ the energy as 1 gram of carbohydrate. A fat molecule has a 2-part structure. The first part is called \_\_\_\_\_\_\_ Attached to the glycerol are 3 long chains called \_\_\_\_\_\_\_

# Formation of Carbohydrates

Photosynthesis (reduction of carbon dioxide) (endothermic reaction)

 $6CO_2 + 6H_2O + energy \rightleftharpoons C_6H_{12}O_6 + 60$ 

Metabolism (oxidation of carbohydrate) (exothermic reaction)





# Lipids:

What do butter, margarine and oil in salad dressing have in common?? They are all lipids. Lipids are organic compounds that and will not dissolve in water (they are insoluble) Fats, oils. waxes, etc., make up this group. Lipids contain the same elements: \_\_\_\_\_,H,O that carbohydrates do, but they are in different proportions. Lipids are a more concentrated source of energy than carbohydrates. They provide twice as much energy per gram as carbohydrates.

#### Saturated vs Unsaturated Fats:

Fats and oils are classified as saturated and unsaturated according to the types of bonds in their carbon chains. \_\_\_\_\_\_\_ such as cheese, whipped cream, ice cream only **contain** \_\_\_\_\_\_\_ between carbons.

contain one or more\_\_\_\_\_

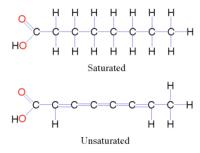
between atoms. Saturated fats: only \_\_\_\_\_\_ bonds in the carbon chain. Most fats "bad" fats. Diets high in saturated fat are linked to heart disease. Unsaturated fats: one or more

"good" fats bonds in the carbon chain. Most oils from

List 2 examples of Saturated Fats:

List 2 examples of Unsaturated Fats:

COOH COOH COOH -COOH



### Cholesterol

Cholesterol is another lipid in cell membranes It is also needed to make like adrenaline

Your body makes the cholesterol that it needs, but it is also found in many foods that come from animals, like meat and eggs. Although you need cholesterol, eating too much of it can block arteries and lead to heart disease.

# **DO BrainPop: Fats** Login as: mms308, marshall

### Video 6: Fueling the Body: Carbohydrates & Fats

- 1. Carbohydrates, stored in the liver & muscles, are converted into what?
- 2. How many hours of energy does the average person have stored?
- 3. Explain what the video meant by "the body cannibalizes itself".

4. Converting fat to fuel requires what?

5. Explain how our ancient ancestors affected the way our bodies work today.

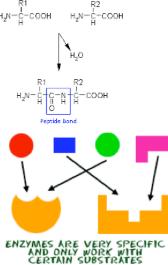
Proteins ( & amino acids)
Proteins are big molecules called \_\_\_\_\_\_ Made of smaller molecules called \_\_\_\_\_\_
Made of carbon, hydrogen, oxygen, nitrogen, sulfur, & some other elements. There are at least \_\_\_\_\_\_
There are \_\_\_\_\_\_ there are \_\_\_\_\_\_\_ proteins in your body. Each has a different \_\_\_\_\_\_\_ that gives it a specific \_\_\_\_\_\_. There are \_\_\_\_\_\_ types of structure, including coils & curls. Milk and fish contain protein, a particular kind of hydrocarbon that is necessary Arrine acid 1 Arrino acid 1 R1 R2 H<sub>2</sub>N-C-COOH H<sub>2</sub>N-C-COOH for all living cells. Proteins are \_\_\_\_\_\_ formed by linking together monomers called \_\_\_\_\_. Think of proteins as being like a word. Amino acids are the letters in that word. Rearranging the letters makes words with different meanings. Proteins are in your muscles, hair, every living cell in your body. of the **amino** acids used by our bodies are absolutely essential for us to function properly, and our bodies DON'T make them. amino acids are made in our bodies, but the other must be obtained by eating \_\_\_\_\_ that contain them. It's important to have lots of protein in

vour diet!



Proteins in foods such as are broken down into amino acids. Without protein, your body can't function perfectly.. This is why it's important for vegetarians to find protein from non-animal sources.

**Enzymes** Some proteins curl up into a shape like a ball of enzymes. An enzyme is a special protein and a \_\_\_\_\_\_ for a chemical reaction in living things. Catalysts speed up the rate of a reaction. Enzymes are needed for many chemical reactions in your body. Without them, these reactions would occur too slowly to keep you alive.



# **Nucleic Acids**

Largest & most complex biomolecule. Includes:

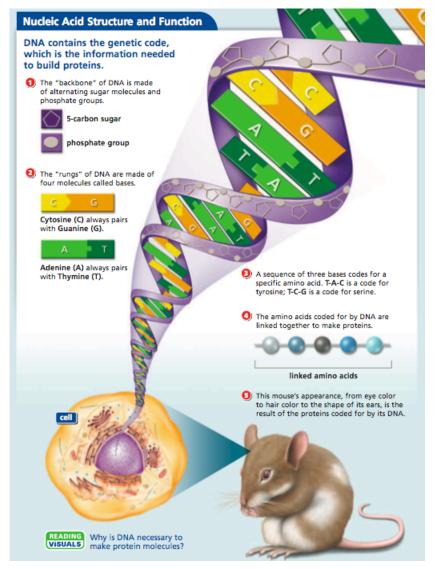
- \_\_\_\_: deoxyribonucleic acid
- \_\_\_\_: ribonucleic acid

Huge, complex carbon-based molecules. Their job: contain information that cells use to make proteins. Made of carbon, hydrogen, oxygen, nitrogen, phosphorous. Every cell in your body has a complete set of nucleic acids. The process of making proteins from amino acids is called

# DNA

Deoxyribonucleic acid. One of the largest molecules. A single DNA molecule has more than \_\_\_\_\_\_atoms. A DNA molecule is a twisted ladder, or a \_\_\_\_\_The sides of the ladder is made of 5-carbon \_\_\_\_\_ molecules called deoxyribose and \_\_\_\_\_ groups. The "rungs" of the ladder are made of: \_\_\_\_\_\_. The 4 Nitrogen bases in a DNA molecule occur in matched sets: \_\_\_\_\_\_(A) to \_\_\_\_\_(T) \_\_\_\_\_(C) to \_\_\_\_\_(G)

The order of the bases in DNA is the way in which DNA stores instructions for making proteins. (A protein is made of amino acids that have to be linked in certain order ) Each of the 20 amino acids is represented by a series of \_\_\_\_\_ DNA bases. **Vitamins:** Most of the chemical needed for life can by made by your own body, like proteins. However, there are certain chemicals that your body does not automatically make. We call these vitamins & \_\_\_\_\_. The only place to get these: food!!



Vitamin C: \_\_\_\_\_\_\_, also called vitamin C, is needed for several important processes in your brain and nervous system. \_\_\_\_\_\_\_ results from a lack of vitamin C in your diet. It causes of spotting on the skin, spongy gums, and bleeding membranes, and can eventually lead to death. The British Royal Navy were among the first to discover this vitamin deficiency, when they noticed their sailors would get sick without fresh fruits & vegetables. Review:

\_\_\_\_\_with other atoms enables it to make a huge 1. Carbon's unique ability to form number of compounds. The structure of its compound determines its properties in the following ways: shorter hydrocarbons are\_\_\_\_\_\_In general, they have low boiling points and the evaporate and burn easier. Longer hydrocarbons\_\_\_\_\_\_\_molecules and exist as solids or liquids at room temp molecules and exist as solids or liquids at room temp 2. Hydrocarbons can be composed of hydrogen and carbon alone, or other chemical groups may be substituted fro hydrogen on the molecule to form new compounds, as with vitamin C. Three additional types of hydroxyl group (-OH) replaces 1 or more hydrogen atoms, produced naturally by sugar fermentation include: in fruit and grain. \_\_\_\_\_: carboxyl (-COOH) replaces a CH3 group, oxygen double bonds with carbon, and produces a sour taste :amine group (-NH3) replaces a hydrogen, found in caffeine, novocaine, and in some vitamins. 3. food providing \_\_\_\_\_: milk, fish, meat, poultry 4. Some examples of biological compounds at work: food providing : pasta, bread, : make up muscles. : found vegetables, sugar in grapes and honey. contain carbohydrates in the form of starch food providing butter, margarine, oils, solid shortening 5. What do they do for our bodies: \_\_\_\_\_\_ provide growth & renewal \_\_\_\_\_\_ provide energy

Brain CARBOHYDRATES	ES	Brain FATS POP	
1 What might happen if you didn't eat enough carbohydrates?	6 Which food contains the most starch?	1 Why is fat an important part of our diet?	6 Why is saturated fat found in so many animal products?
A Your body wouldn't be able to build muscle	A Pasta	A Because it tastes good	A Because it's part of the cell membranes of animal tissues
B Your body wouldn't have enough energy to function	B Broccoli	B Because it carries so much energy	B Because animals must eat saturated fat in order to survive
C You'd develop serious vitamin deficiencies	C Table sugar	C Because it contains nutrients you can't get from anything else	C Because proteins are broken down into saturated fat when animals die
D Your bones and teeth would become weaker	D Steak	D Because it contains oxygen atoms	D Because saturated fat is a key component of animal bones
2 Carbohydrates are to the human body as what is to a car?	7 Where in your body are complex carbohydrates broken down into simple sugars?	2 What do carbohydrates and fats have in common?	7 If you eat too much saturated fat, cholesterol might build up:
A Transmission fluid	A In your bloodstream	A They're both sources of glucose	A In your brain
B Antifreeze	B In your intestines	B They're both sources of protein	B In your liver
C Mator oil	C In your mouth	C They're both sources of fatty acids	<b>C</b> In your veins and arteries
D Gasoline	D In your liver	D They're both sources of glycerol	D In your bones
3 What suffix indicates that a chemical is a sugar?	8 If you're hungry, but don't want your blood sugar to spike, what should you eat?	3 What might happen if you didn't get enough fatty acids in your diet?	8 What is a key difference between saturated and unsaturated fats?
A ose	A A baked potato		A Saturated fats are usually liquid at room temperature; unsaturated fats are usually solid
B ase	B An apple		B saturated lats stay solid at room temperature; unsaturated lats are usually liquid
C -ate	C Sushi with white rice	C Your kidneys and liver might not function properly	
D -ite	D A candy bar	D Your bones might become brittle	D Saturated fats are healthier than unsaturated fats
4 Which of the following contains complex carbohydrates?	<b>9</b> Which of the following breads is the healthiest?	4 A sedentary lifestyle can cause glucose to turn into body fat. What is the best synonym for "sedentary"?	9 A product that contains lots of hydrogenated oils is probably rich in:
A	A White	A Healthy	A Saturated fat
1	B Italian	B Hyperactive	B Unsaturated fat
	C Whole wheat	C Sleep-deprived	C Monosaturated fat
	D Sourdough	D Inactive	D Trans fat
5 How is fiber different from most other foods you eat?	*	5 Which of the following items might be high in saturated fat?	10 Why shouldn't you worry about getting enough fat in your diet?
A It doesn't contain any sugars at all	A Fructose	A	A Because all fat is bad for you
	R Sucrose	B	B Because it's found in so many foods
C Your body can't digest it	C Glucose	°	C Because you don't need any fat at all in your diet
D It contains elements of all the major food groups	D Starch		because you can take it as a vitamin it you don't eat it

**QUIZ** YOURSELF What is the difference between sugar, starch, and cellulose?

List 3 differences between saturated and unsaturated fats:

**<u>OUIZ YOURSELF</u>**: How are lipids different from carbohydrates?

I	
Water is the most abundant molecule in the human body. ( What does "most abundant" mean?	6 What might happen if you consumed too many lipids?
Most useful	A You might develop heart disease
Most common	B Your cells might swell up with too much water
Smallest	C You might get a very high fever
Simplest	D You might wind up with too much energy
What might happen if you did not consume carbohydrates?	if you did not consume carbohydrates? $ {f 7}$ In your body, where can you find protein?
Your body would dry out	A Only in your hair and nails
You would die from a lack of oxygen	B Only within your bones
You wouldn't get the energy you needed to function	C Only in tissues and cartilage
Your body wouldn't be able to build new muscle mass	D Just about everywhere
The four elements that make up 95 percent of the body's { weight are carbon, hydrogen, oxygen, and:	8 What are nucleic acids responsible for?
Nitrogen	
Nickel	
Calcium	D Delivering chemical messages between the brain and the body
If the human body were a car, glucose would be:	2 B.I
DI	
The wheels	
	D The reaction with the enzyme would require less energy
What might happen if you didn't consume enough lipids? 1	10 Why should you eat right?
You would die of dehydration	A Because it'll make you thin and good-looking
Your body might not be able to store energy	B Because your doctor and parents say so
You might not be able to go to the bathroom	C Because your body will gradually lose water if you don't
Your cells would not be able to break down sugars	D To provide your body with the proper raw materials it needs to function
	ater is the most abundant" mean?       in the human body.         Most useful         Most common         Simplest         Your body would dry out         You wouldn't get the energy you needed to function         Your body would in the able to build new muscle mass         Perform elements that make up 95 percent of the body's feight are carbon, hydrogen, oxygen, and:         Your body would n't be able to build new muscle mass         Perform elements that make up 95 percent of the body's feight are carbon, hydrogen, oxygen, and:         Your body wore a car, glucose would be:         Nitrogen         Nickel         Iron         Calcium         The human body were a car, glucose would be:         The mator oil         The wheels         The gasoline         Nat might not be able to store energy         Your body might not be able to store energy         Your would die of dehydration         Your would not be able to break down sugars.